QUANTITATIVE STUDY OF THE INTERACTION OF AMINOACYL-tRNA WITH THE A SITE OF ESCHERICHIA COLI RIBOSOMES

Equilibrium and kinetic parameters of binding in the absence of EF-Tu factor and GTP

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1. Introduction

Phe-tRNA^{Phe} has a higher affinity to the D site of 70 S ribosomes over wide concentrations of Mg²⁺ and in the absence of elongation factor Tu (EF-Tu) and GTP [1]. If ribosomes were in excess, Phe-tRNA^{Phe} was found exclusively at the D site. This result allowed us to derive the thermodynamic parameters of the D-site binding of aminoacyl-tRNA [2].

On the other hand, if high concentrations of PhetRNA^{Phe} were used, the A site began to fill after D-site saturation, and quantitative formation of (Phe)₂-tRNA^{Phe} occurred [3]. It made impossible the determination of thermodynamic parameters of A-site binding.

Here, to avoid this difficulty, we used 70 S ribosomes, where the D sites were initially preoccupied by an analogue of peptidyl-tRNA, N-acetyl-[14 C]PhetRNA $_{\rm OX,-red.}^{\rm Phe}$, exempt from donor activity [4–6]. In this system we measured both association and dissociation rate constants, and the equilibrium binding constants of Phe-tRNAPhe with the vacant A site of 70 S · poly(U) · peptidyl-tRNAPhe ox.-red. complex. The latter appeared by 2–3 orders of magnitude lower, than the binding constant of Phe-tRNAPhe with the D site of 70 S · poly(U) complex.

2. Materials and methods

30 S and 50 S ribosomal subunits were isolated from *Escherichia coli* MRE-600 as in [3]. Enriched preparations of [¹⁴C]Phe-tRNA^{Phe} (1440 pmol/A₂₆₀ unit) and [³H]Phe-tRNA^{Phe} (1440 pmol/A₂₆₀ unit),

as well as fractionated poly(U) (30 000 $M_{\rm r}$) were prepared as in [8,9]. Enriched deacylated tRNAPhe was obtained by deacylation of a labeled preparation Phe-tRNAPhe. Its oxidation—reduction was performed according to [4,10]. To remove the traces of Na-borohydride, the final preparation of tRNAPhe ox.—red. was passed additionally through a Sephadex G-15 column in 0.05 M Na-acetate buffer (pH 5.5). Aminoacylation of tRNAPhe was made in 1 ml buffer: 0.1 M Tris—HCl (pH 7.6); 0.02 M MgCl₂; 0.075 M NH₄Cl; 0.006 M 2-mercaptoethanol, containing 20 A_{260} units tRNAPhe ox.—red. 100 nmol [14C]phenylalanine, 4 μ mol ATP and 120 μ g S-100 supernatant free of tRNAs. After 60 min incubation at 37°C aminoacylation yield attained 620 pmol [14C]Phe/ A_{260} unit. Acetylation of [14C]Phe-tRNAPhe ox.—red. was performed according to [11]; the yield of acetyl-Phe-tRNAox.—red. was usually 97—98%. Then N-acetyl-[14C]Phe-tRNAPhe ox.—red. was additionally purified on a BD-cellulose column to a final enrichment 1350 pmol/ A_{260} unit.

To obtain S-100 supernatant free of tRNAs, the supernatant fraction (after the first pelleting of ribosomes from the crude extract [3]) was dialysed against buffer A: 0.02 M Tris—HCl (pH 7.3); 0.01 M MgCl₂; 0.3 M NaCl; 0.5 mM EDTA; 6 mM 2-mercaptoethanol. Then the protein was applied to a DEAE-cellulose column (3000 A_{280} units per 40 \times 3.6 cm column) in buffer A, and eluted with the same buffer. Under these conditions tRNAs were adsorbed on DEAE-cellulose, and A_{280}/A_{260} ratio of the final S-100 supernatant was 1.5 instead of initial 0.7.

The binding assays were performed in the buffer I: 0.02 M Tris—HCl (pH 7.3); 0.2 M NH₄Cl, containing different concentrations of Mg²⁺ (as indicated in

figure legends). Reaction with puromycin was as in [12]. For double-label counting (14 C/ 3 H) nitrocellulose filters were dissolved in 1 ml acetone followed by addition of 15 ml scintillation liquid (3 g PPO + 0.1 g POPOP per liter toluene).

3. Results and discussion

It was shown earlier that peptidyl-tRNA binds exclusively at the D site of 70 S · poly(U), if the latter is in excess; the ratio of association constants of peptidyl-tRNA with the D and A sites, K_a^D/K_a^A , being equal to 30–50 [7,13]. As can be seen from table 1, N-acetyl-[14C]Phe-tRNAPhe ox.—red., in these conditions, binds also exclusively at the D site, because the binding is not sensitive to tetracycline. The lack of puromycin reactivity proves the quantitative cleavage of the C2'-C3' bond in 3'-terminal adenosine of acetyl-Phe-tRNAPhe ox.—red. (compare lines 2 and 4, table 1). Fig.1A shows that modified acetyl-Phe-tRNAPhe ox.—red. binds to 70 S ribosomes as fast as the unmodified one [7]; moreover, the association constants of both analogues of peptidyl-tRNA are indistinguishable ((2.8 \pm 0.2) \times 108 M⁻¹ at 8 mM Mg²⁺, fig.1B).

To saturate the D sites quantitatively at various conditions, we measured K_a^D values for N-acetyl-[14C]Phe-tRNAPhe at different Mg²+ concentrations and temperatures (table 2). From these data the enthalpy of interaction was calculated at 8 mM Mg²+ (13 ± 2 kcal/mol); it appeared close to the value found earlier for the system with unmodified N-acetyl-[14C]Phe-tRNAPhe [7].

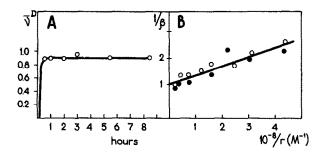


Fig.1. Binding of N-acetyl-[14C]Phe-tRNA $_{\text{OX.-red.}}^{\text{Phe}}$ at the D site of 70 S · poly(U) complex at 0°C. (A) Kinetics of binding. Incubation mixtures contained in 100 μ l buffer I (with 10 mM Mg²⁺): 6 pmol 30 S subunits; 9 pmol 50 S subunits; 5 μ g poly(U); 10 pmol N-acetyl-[14C]Phe-tRNA $_{\text{OX.-red.}}^{\text{Phe}}$; $_{\nu}^{\text{D}}$), no. acetyl-Phe-tRNA molecules bound/ ribosome. (B) Adsorption isotherms of N-acetyl-[14C]Phe-tRNA $_{\text{OX.-red.}}^{\text{Phe}}$ (-•-) and N-acetyl-[14C]Phe-tRNA $_{\text{OX.-red.}}^{\text{Phe}}$ (-•-) at the D site of 70 S · poly(U) complex, measured by the method of variable volume [1]. Incubation mixtures contained in 0.1-4.0 ml buffer I (with 10 mM Mg²⁺): 9 pmol 30 S subunits; 12 pmol 50 S subunits; 15 μ g poly(U); 5 pmol peptidyl-tRNA; incubation, 2 h at 0°C, β , portion of peptidyl-tRNA bound to ribosomes at equilibrium; r, concentration of free 70 S · poly(U) complex.

All following experiments were in two steps:

- (i) The ternary 70 S · poly(U) · peptidyl-tRNA Phe complex was formed with controlled $\overline{\nu}^D$ values = 1.0 ± 0.1 (as in fig.2A.3).
- (ii) After attainment of equilibrium, [3H]PhetRNA^{Phe} was added.

Fig.2A demonstrates that the non-enzymatic binding of aminoacyl-tRNA at the A site is very slow; several hours are needed to reach equilibrium at 0°C. This

Table 1
Reaction of prebound N-acetyl-[14C]Phe-tRNAPhe and N-acetyl-[14C]Phe-tRNAPhe with puromycin

Expt.	tRNA	Tetra- cycline (2 × 10 ⁻⁵ M)	tRNA bound (pmol)	Portion tRNA bound (β)	N-Acetyl-Phe-puro- mycin synthesized	
					(pmol)	(%)
1	N-Acetyl-	+	4.47	0.99	0	0
2	Phe-tRNAPhe oxred.	-	4.56	1.01	0	0
3	N-acetyl-	+	3.42	0.76	2.85	84
4	Phe-tRNA Phe	energy.	3.77	0.84	2.95	78

Incubation mixtures contained in 100 μ l buffer I (with 20 mM Mg²+): 10 pmol 30 S subunits; 15 pmol 50 S subunits; 5 μ g poly(U); 4.5 pmol N-acetyl-[¹⁴C]Phe-tRNAPhe or N-acetyl-[¹

Table 2
K _a ^D values for N-acetyl-[14C]Phe-tRNAPhe _{OXred.} at different Mg ²⁺
concentrations and temperatures
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Mg ²⁺ (mM)	$K_{\rm a}^{\rm D} \times 10^{-8} \; ({\rm M}^{-1})$					
	0°C	10°C	19°C	32°C		
4	0.74 ± 0.06					
6	2.1 ± 0.1					
6 8	2.8 ± 0.2	2.6 ± 0.1	0.72 ± 0.05	0.26 ± 0.06		
10	~10			0.56 ± 0.15		
15	>10			4.2 ± 0.5		
20	>10			4.4 ± 0.4		

Measurements were made as described in legend to fig.1B

binding is reversible, because after dilution of the incubation mixtures [${}^{3}H$]Phe-tRNAPhe dissociates from the complex, and $\overline{\nu}^{A}$ value tends to a new equilibrium value (cf. $-\Box$ and $-\triangle$, fig.2A). It is essential that $\overline{\nu}^{D}$ does not change upon dilution at $0^{\circ}C(-\blacksquare$, $-\triangle$, fig.2A). These data provide us criteria for the correct determination of equilibrium K_{a}^{A} constants.

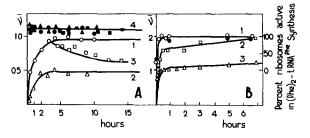


Fig. 2. (A) Kinetics of association and dissociation of [³H]PhetRNA Phe at the A site of ribosomes at 0°C. Two sets of incubation mixtures contained in buffer I (with 10 mM Mg²+): 5 pmol 30 S subunits; 7 pmol 50 S subunits; 5 μ g poly(U); 11 pmol N-acetyl-[¹⁴C]Phe-tRNA Phe mixture vol. 100 μ l; 40 pmol N-acetyl-[¹⁴C]Phe-tRNA Phe $\phi_{X.-red.}$, mixture vol. 600 μ l. After 2 h incubation at 0°C 11 pmol [³H]PhetRNA Phe was added to each mixture, and kinetics of aminoacyl-tRNA binding was measured both in more (-0-) and less (-\(^{\text{\text{0}}}\)-) concentrated mixtures. After 3 h 30 pmol N-acetyl-[¹⁴C]Phe-tRNA Phe $\phi_{X.-red.}$ was added to the part of more concentrated mixtures followed by dilution (with pure buffer I) to 600 μ l (-\(^{\text{0}}\)-). Curve 4, control. $\overline{\nu}^{\text{D}}$ values for N-acetyl-[¹⁴C]Phe-tRNA Phe $\phi_{X.-red.}$ in expt: 1 (-\(^{\text{\text{0}}}\)-); 2 (-\(^{\text{\text{\text{0}}}}\)-); 3 (-\(^{\text{\text{\text{0}}}}\)-); B) Kinetics of binding of [¹⁴C]Phe-tRNA Phe and formation of ([¹⁴C]Phe)₂-tRNA Phe. Incubation mixtures contained in 200 μ l buffer I (with 20 mM Mg²+): 10 pmol 30 S subunits; 15 pmol 50 S subunits; 5 μ g poly(U); 58 pmol (-\(^{\text{0}}\)-, curve 1), 34 pmol (-\(^{\text{\text{0}}}\)-, curve 2), 20 pmol (-\(^{\text{\text{0}}}\)-, curve 3) [¹⁴C]Phe-tRNA Phe. In one of these experiments the fraction of ribosomes active in ([¹⁴C]Phe)₂-tRNA Phe formation was determined, as in [3].

Results of these measurements are documented in fig.3 and table 3. We see that association constants of codon-dependent, factor-free binding of Phe-tRNA Phe with the A site of the ternary complex appear equal to $2.4 \times 10^7~\text{M}^{-1}$ at 10~mM and $6 \times 10^7~\text{M}^{-1}$ at 20~mM Mg²+ and 0°C. After comparison of these values with those for the D-site binding ($K_a^D = 2 \times 10^9~\text{M}^{-1}$ at 7 mM and $\sim 10^{10}~\text{M}^{-1}$ at 10 mM Mg²+ and 0°C [1,2]), we draw the conclusion that the affinity of Phe-tRNA Phe to the D site is by 2–3 orders of magnitude higher, than to the A site.

The very slow kinetics of Phe-tRNA^{Phe} binding at the A site is not the consequence of the presence of artificial peptidyl-tRNA at the D site. The same situation we observe when intact [¹⁴C]Phe-tRNA^{Phe} fills both D and A sites (fig.2B). The first molecule of

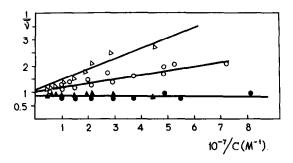


Fig. 3. Adsorption isotherms of [3 H]Phe-tRNAPhe at the A site of ribosomes at 0°C. Incubation mixtures contained in 150 μ l buffer I with different Mg²⁺ concentrations: 7 pmol 30 S subunits; 10 pmol 50 S subunits; 5 μ g poly(U); 13 pmol N-acetyl-[14 C]Phe-tRNAPhe was added to each mixture and incubation was continued for 3 h. Mg²⁺ was 10 mM ($^{-\Delta}$) and 20 mM ($^{-\Delta}$). Control: D values for N-acetyl-[14 C]Phe-tRNAPhe ($^{-\Delta}$) in both experiments ($^{-\Delta}$ and $^{-\Phi}$, respectively).

Table 3	
KA values for [3H]Phe-tRNAPhe at 0°C and different Mg2+ c	concentrations

	8 mM Mg ²⁺	10 mM Mg ²⁺	15 mM Mg ²⁺	20 mM Mg ²⁺	30 mM Mg ²⁺
$K_{\rm a}^{\rm A} \times 10^{-7} \ ({\rm M}^{-1})$	1.87 ± 0.24	2.40 ± 0.30	5.0 ± 1.0	6.0 ± 1.0	8.3 ± 1.2

Measurements were made as described in legend to fig.3

aminoacyl-tRNA binds very fast, obviously, at the D site, because the binding is not inhibited by tetracycline (not shown). The A site is occupied much slower ($\neg\neg$, $\neg\triangle$, at comparable concentrations of Phe-tRNAPhe, fig.2A); this binding is not reversible, because (Phe)₂-tRNAPhe is quantitatively formed while filling the A site ($-\bullet$, fig.2B).

Kinetic data in fig.2A allow us to compute the rate constants of association and dissociation of [³H]Phe-tRNA^{Phe}, if the process is presumably described by the simple scheme:

$$T + A \stackrel{k_1}{\rightleftharpoons} TA \tag{1}$$

where T is Phe-tRNAPhe, A the vacant A site of the ternary 70 S · poly(U) · acetyl-Phe-tRNAPhe ox.—red. complex, k_1 association and k_{-1} dissociation rate constants. The mean value of k_1 (determined from —0—, $-\triangle$ —, fig.2A), is equal to $(2\pm1)\times 10^5$ l/mol × min. As far as the equilibrium constant, K_a^A , is known (see table 3), it is easy to compute $k_{-1} = k_1/K_a^A \simeq 8\times 10^{-3}$ min⁻¹. On the other hand, k_{-1} can be estimated independently from the experiment on dissociation of Phe-tRNAPhe (—0—, fig.2A). Surprisingly, in this case k_{-1} appears equal to 1.2×10^{-3} min⁻¹, i.e., the dissociation of Phe-tRNAPhe occurs much slower than expected.

Repeated experiments confirmed this discrepancy. This result led us to suggest, that the non-enzymatic interaction of aminoacyl-tRNA with the A site is more complicated and may be described, for instance, by the scheme:

$$T + A \xrightarrow{k_1} TA^* \xrightarrow{k_2} TA$$
 (2)

Here, TA* is an intermediate which is formed at first converting slowly into final TA complex, where Phe-tRNA^{Phe} becomes competent in its acceptor

activity. At the first step of binding, quasi-equilibrium is attained, the admixture of TA complex is negligible and the binding is described by eq. (1). But in longer dissociation experiments (——, fig.2A) TA complex is appreciably accumulated. Such sequence of events (initial location of aminoacyl-tRNA at an intermediate site followed by its fastening at a functional A site) explains why the real rate constant of dissociation is lower than the expected one.

If so, we find some common features between the factor-free and factor-dependent binding of aminoacyl-tRNA at the A site. EF-Tu-dependent binding occurs, at least in two-steps. The first step. [GMP-P(CH₂)P], can be easily distinguished because peptidyl transfer cannot occur in the presence of nonhydrolyzable analogue of GTP [14]. However, after hydrolysis of GTP and removal of the EF-Tu · GDP complex from ribosome, aminoacyl-tRNA becomes competent in peptide bond formation. These findings have led to the assumption [15] of the 'accomodation' of aminoacyl-tRNA within the ribosome, and to the intermediate R (recognition) site, which is distinct from the A site [16]. At least two-step binding is needed for the selection process [17]. It is likely, that both factor-free and factor-promoted binding of aminoacyl-tRNA at the ribosomal A site occur in a similar way, the association rates of both steps being strongly increased by EF-Tu + GTP.

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